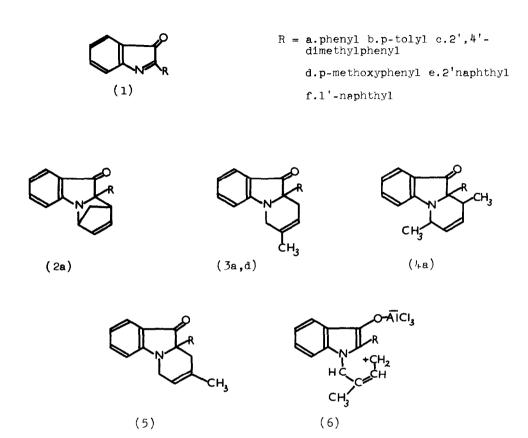
THE REACTION OF INDOLONES (3-OXOINDOLENINES) WITH DIENES. H.S. Ch'ng and M. Hooper

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We have recently prepared the previously unknown 2-phenylindolones with electron releasing substituents in the phenyl ring (lb-d) and 2-naphthyl indolones (le-f) from the corresponding indoles by the well documented method of Baeyer.^{1,2*} All the reactions proceeded smoothly in high yields (80-90%). The products were isolated as stable red solids and identified by infrared, mass spectra, $\frac{3}{2}$ and elemental analysis. The indolones (la,b) reacted at room temperature with cyclopentadiene in benzene solution and with 2-methylbutadiene and hexa-2,4-diene in refluxing benzene in the presence of aluminium trichloride or perchloric acid giving the bright yellow Diels-Alder adducts (2a). (3a,d), and (4a). The adducts were identified by infrared spectroscopy (nujol) 1690 (C=O), 1655 (C=C) cm.⁻¹, mass spectrometry and elemental analysis. The mass spectrum of the adduct (2a) showed a parent ion m/e 273 and fragment ions at 208(M-65)⁺ and 207(M-66)⁺ as well as the expected ions resulting from the breakdown of 2-phenylindolone.³ The structure of the adduct formed by the reaction of (1a) with isoprene may be (3a) or (5); structure (3a) is favoured since the ionic intermediate (6) leading to it would have the lowest energy.⁴ Diels-Alder reactions in which the azomethine group functions as a dienophile are rare.⁵ Under the above conditions only starting material was recovered from the reaction of (1a) or (1b) with butadiene, 1.3cyclohexadiene, furan, and anthracene. Although the azomethine linkage of indolones is highly reactive⁶ the very limited extent of their reactions with dienes provides further evidence for the reluctant participation of this grouping in the Diels-Alder reaction.

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